

AFCRL-66-545

AUGUST 1966

PHYSICAL SCIENCES RESEARCH PAPERS, NO. 255

AD 638826

UPPER ATMOSPHERE PHYSICS LABORATORY PROJECT 8627

AIR FORCE CAMBRIDGE RESEARCH LABORATORIES

L. G. HANSCOM FIELD, BEDFORD, MASSACHUSETTS

Electron Capture from Atomic Nitrogen by Protons

ROBERT A. MAPLETON

Reprinted from THE PHYSICAL REVIEW, Vol. 145, No. 1, pp. 25-26, 6 May 1966

Distribution of this document is unlimited

OFFICE OF AEROSPACE RESEARCH
United States Air Force

OFFICE OF AEROSPACE RESEARCH

Electron Capture from Atomic Nitrogen by Protons

ROBERT A. MAPLETON

Air Force Cambridge Research Laboratories, Bedford, Massachusetts

(Received 17 December 1965)

The total Oppenheimer-Brinkman-Kramers (OBK) cross sections for 1s-, 2s-, and 2p-orbital electron capture from atomic nitrogen by protons are adjusted to correspond to capture from N₂ by deuterons, and are compared with recent measurements of Berkner *et al.* Calculated cross sections for 1s capture are dominant above 3 MeV, and the total OBK cross section still exceeds the measured cross section at the deuteron energy of 21.5 MeV.

RECENTLY published measurements of cross sections for electron capture from N₂ by deuterons differ markedly from the OBK (Oppenheimer-Brinkman-Kramers) cross sections for electron capture from N by protons when expressed as cross sections per gas molecule at the appropriate impact energy for deuterons.¹ One of the chief sources of the disagreement in this comparison originates from using calculated cross sections for only p-orbital capture at impact energies where capture from inner subshells is dominant.²

The importance of 1s- and 2s-orbital capture at high impact energies was emphasized in I, but only estimates of these cross sections could be given since the wave functions for the atomic ions, N^{+(4S, 3S)}, configurations (1s)²2s(2p)³ and 1s(2s)²(2p)³, were not available during the time that the calculations were effected. Since wave functions have become available,³ OBK cross sections for 2s-orbital capture have been computed and published.⁴ Although wave functions for the configuration 1s(2s)²(2p)³ have not been published, the atomic orbitals for the ion Ne⁺, configurations 1s(2s)³(2p)⁶ and (1s)²2s(2p)⁶, are given in a recent paper.⁵ A comparison of the orbital functions for these two configurations of Ne⁺ shows that the two sets of 1s and 2s parameters differ very little. With this in mind, it is assumed that not much error in the OBK cross sections for 1s capture result from using Roothaan³ atomic orbitals of (1s)²2s(2p)³, N^{+(4S, 3S)}}, to represent the corresponding term values of 1s(2s)³(2p)³. (This is a very special application of the wave functions, and is not supposed to imply the validity of such an approximation for other purposes.)}

This approximation has been used to compute cross sections for 1s capture, impact energies ≥ 1 MeV; moreover, the energy range of the cross sections for 2s capture⁴ has been extended to 100 MeV. The cross sections are expressed as a function of the impact energy of the proton in the frame of reference where the atomic target is initially at rest. Perhaps the most notable distinction of these OBK cross sections is the

dominance of 1s-orbital capture for impact energies exceeding 3 MeV. These cross sections pertain to capture into the 1s state of atomic hydrogen only, and the n^{-3} rule is used to sum the cross sections into all s states of hydrogen.⁶ As for the contributions of simultaneous charge transfer and excitation, previous calculations for helium suggest that these processes can be neglected with small error.⁷ The sum of the OBK cross sections for 1s-, 2s-, and 2p-orbital capture from N^(4S) into all s states of the hydrogen atom leaving the residual ions N^{+[3P; (1s)²(2s)²(2p)²]}, N^{+[3S, 3S; (1s)²2s(2p)³]}, and N^{+[3S, 3S; 1s(2s)²(2p)³]} have been multiplied by 2, and are plotted in Fig. 1.

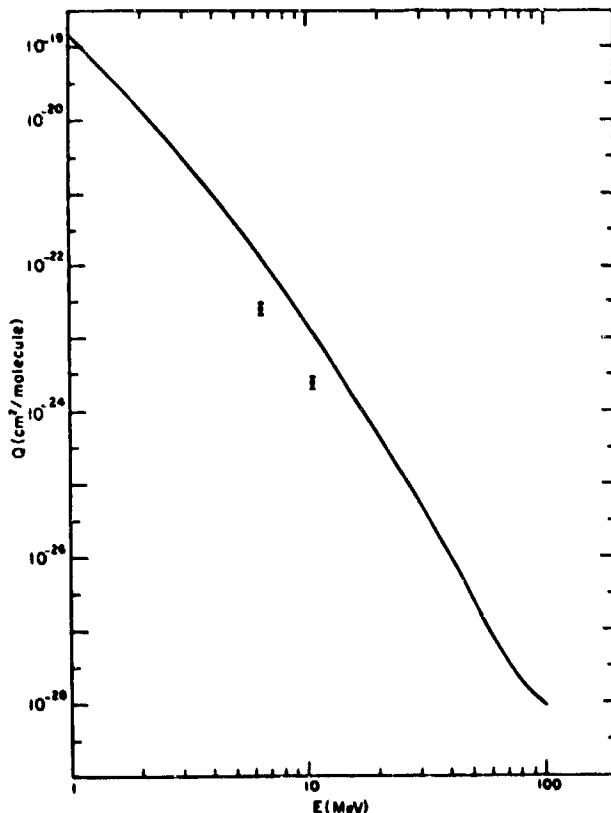


FIG. 1. Electron-capture cross sections per N₂ molecule. E is the impact energy of the proton in MeV. Closed circles with error bar represent experimental values taken from Table I of Berkner *et al.* Solid curve represents the sum of the OBK cross sections for atomic nitrogen multiplied by 2. (See text for the processes represented.)

¹ K. H. Berkner, S. N. Kaplan, G. A. Paulikas, and R. V. Pyle, Phys. Rev. 140, A729 (1965).

² R. A. Mapleton, Phys. Rev. 130, 1829 (1963). This is referred to as I.

³ C. C. J. Roothaan and P. S. Kelley, Phys. Rev. 131, 1177 (1963).

⁴ R. A. Mapleton, Proc. Phys. Soc. (London) 85, 1109 (1965).

⁵ P. S. Bagus, Phys. Rev. 139, A619 (1965).

⁶ J. R. Oppenheimer, Phys. Rev. 31, 349 (1928); J. D. Jackson and H. Schiff, *ibid.* 89, 359 (1953).

⁷ R. A. Mapleton, Phys. Rev. 122, 528 (1961).

It is seen that the two experimental values of Berkner, *et al.* are less than the corresponding OBK values, which fact suggests that the asymptotic energy region may not yet be reached. Of course, it is not known whether OBK cross sections are the asymptotic values of a correct theory, and at what energy the onset of the asymptotic value would occur. Little is also known how well electron capture from N₂ can be described in terms of N atoms⁸ (factor of 2) or how much the present OBK values would be altered by a recalculation with improved wave functions. In the author's

opinion, it is very difficult to decide theoretically what the asymptotic cross section is for a target as complicated as N₂.

An atomic system much more tractable to theoretical analysis is helium, and the energy range of the OBK calculations⁹ using the 6-parameter helium wave function have also been extended to 100 MeV. The OBK cross sections for capture from He, N, and O, described in this paper, can be obtained from the author.

Gratitude is expressed to Professor A. Dalgarno for informing the author of these measurements.

⁸T. F. Tuan and E. Gerjuoy, Phys. Rev. 117, 756 (1960).

⁹R. A. Mapleton, Phys. Rev. 130, 1839 (1963).

Unclassified
Security Classification

DOCUMENT CONTROL DATA - R&D		
(Security classification of title, body of abstract and indexing annotation must be entered when the overall report is classified)		
1. ORIGINATING ACTIVITY (Corporate author) Hq AFCRL, OAR (CRU) United States Air Force Bedford, Massachusetts		2a. REPORT SECURITY CLASSIFICATION Unclassified 2d. GROUP
1. REPORT TITLE Electron Capture from Atomic Nitrogen by Protons		
4. DESCRIPTIVE NOTES (Type of report and inclusive dates) Journal Article. Interim.		
5. AUTHOR(S) (Last name, first name, initial) MAPLETON, Robert A.		
6. REPORT DATE August 1966	7a. TOTAL NO. OF PAGES 8	7d. NO. OF REFS 9
8a. CONTRACT OR GRANT NO.	8a. ORIGINATOR'S REPORT NUMBER(S) AFCRL-66-545 PSRP No. 255	
b. PROJECT AND TASK NO. 8627-06	9d. OTHER REPORT NO(S) (Any other numbers that may be assigned this report) AFCRL-66-545	
c. DOD ELEMENT 61445014		
d. DOD SUBELEMENT 681310		
10. AVAILABILITY/LIMITATION NOTICES Distribution of this document is unlimited.		
11. SUPPLEMENTARY NOTES Reprinted from the Physical Review, Vol. 145, No. 1, pp. 25-26, 8 May 1966	12 SPONSORING MILITARY ACTIVITY Hq AFCRL, OAR (CRD) United States Air Force Bedford, Massachusetts	
13. ABSTRACT <p>The total Oppenheimer-Brinkman-Kramers (OBK) cross sections for 1s-, 2s-, and 2p-orbital electron capture from atomic nitrogen by protons are adjusted to correspond to capture from N₂ by deuterons, and are compared with recent measurements of Berkner et al. Calculated cross sections for 1s capture are dominant above 3 MeV, and the total OBK cross section still exceeds the measured cross section at the deuteron energy of 21.5 MeV.</p>		

DD FORM 1 JAN 64 1473

Unclassified
Security Classification

Unclassified
Security Classification

14. KEY WORDS	LINK A		LINK B		LINK C	
	ROLE	WT	ROLE	WT	ROLE	WT
1s-orbital capture electronic capture 2s-, 2p-orbital capture Atomic nitrogen						
INSTRUCTIONS						
1. ORIGINATING ACTIVITY: Enter the name and address of the contractor, subcontractor, grantee, Department of Defense activity or other organization (<i>corporate author</i>) issuing the report.	10. AVAILABILITY/LIMITATION NOTICES: Enter any limitations on further dissemination of the report, other than those imposed by security classification, using standard statements such as:					
2a. REPORT SECURITY CLASSIFICATION: Enter the overall security classification of the report. Indicate whether "Restricted Data" is included. Marking is to be in accordance with appropriate security regulations.	(1) "Qualified requesters may obtain copies of this report from DDC."					
2b. GROUP: Automatic downgrading is specified in DoD Directive 5200.10 and Armed Forces Industrial Manual. Enter the group number. Also, when applicable, show that optional markings have been used for Group 3 and Group 4 as authorized.	(2) "Foreign announcement and dissemination of this report by DDC is not authorized."					
3. REPORT TITLE: Enter the complete report title in all capital letters. Titles in all cases should be unclassified. If a meaningful title cannot be selected without classification, show title classification in all capitals in parenthesis immediately following the title.	(3) "U. S. Government agencies may obtain copies of this report directly from DDC. Other qualified DDC users shall request through .."					
4. DESCRIPTIVE NOTES: If appropriate, enter the type of report, e.g., interim, progress, summary, annual, or final. Give the inclusive dates when a specific reporting period is covered.	(4) "U. S. military agencies may obtain copies of this report directly from DDC. Other qualified users shall request through .."					
5. AUTHOR(S): Enter the name(s) of author(s) as shown on or in the report. Enter last name, first name, middle initial. If military, show rank and branch of service. The name of the principal author is an absolute minimum requirement.	(5) "All distribution of this report is controlled. Qualified DDC users shall request through .."					
6. REPORT DATE: Enter the date of the report as day, month, year, or month, year. If more than one date appears on the report, use date of publication.	If the report has been furnished to the Office of Technical Services, Department of Commerce, for sale to the public, indicate this fact and enter the price, if known.					
7a. TOTAL NUMBER OF PAGES: The total page count should follow normal pagination procedures, i.e., enter the number of pages containing information.	11. SUPPLEMENTARY NOTES: Use for additional explanatory notes.					
7b. NUMBER OF REFERENCES: Enter the total number of references cited in the report.	12. SPONSORING MILITARY ACTIVITY: Enter the name of the departmental project office or laboratory sponsoring (paying for) the research and development. Include address.					
8a. CONTRACT OR GRANT NUMBER: If appropriate, enter the applicable number of the contract or grant under which the report was written.	13. ABSTRACT: Enter an abstract giving a brief and factual summary of the document indicative of the report, even though it may also appear elsewhere in the body of the technical report. If additional space is required, a continuation sheet shall be attached.					
8b, 8c, & 8d. PROJECT NUMBER: Enter the appropriate military department identification, such as project number, subproject number, system numbers, task number, etc.	It is highly desirable that the abstract of classified reports be unclassified. Each paragraph of the abstract shall end with an indication of the military security classification of the information in the paragraph, represented as (TS), (S), (C), or (U).					
9a. ORIGINATOR'S REPORT NUMBER(S): Enter the official report number by which the document will be identified and controlled by the originating activity. This number must be unique to this report.	There is no limitation on the length of the abstract. However, the suggested length is from 150 to 225 words.					
9b. OTHER REPORT NUMBER(S): If the report has been assigned any other report numbers (either by the originator or by the sponsor), also enter this number(s).	14. KEY WORDS: Key words are technically meaningful terms or short phrases that characterize a report and may be used as index entries for cataloging the report. Key words must be selected so that no security classification is required. Identifiers, such as equipment model designation, trade name, military project code name, geographic location, may be used as key words but will be followed by an indication of technical context. The assignment of links, rules, and weights is optional.					

Unclassified
Security Classification